

Homework Phy 4523

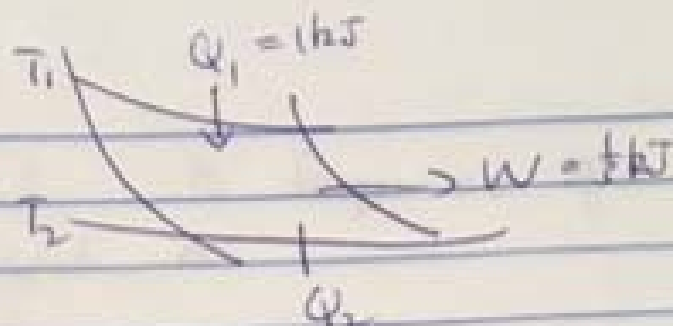
Chap. 2

5.

$$\frac{Q_2}{T_2} = \frac{Q_1}{T_1}$$

$$\therefore Q_2 = Q_1 \frac{T_2}{T_1} = \frac{2}{3} \text{ kJ}$$

$$\therefore W = \frac{1}{3} \text{ kJ}$$



8(a)

Bath
 T_B
 373K

System
 A
 T_A
 273K

$\alpha = \frac{T_B}{T_A}$

$$\Delta S_{\text{System}} = \int_{273}^{323} \frac{dQ}{T} = c \ln \frac{323}{273} = 0.312 \text{ C}$$

$$\Delta S_{\text{Bath}} = \frac{-100 \text{ C}}{373} = -0.268 \text{ C}$$

$$\Delta S_{\text{univ}} = 0.044 \text{ C}$$

(b)

$$\Delta S_{\text{system}} = \int_{373}^{523} \frac{c dT}{T} + \int_{523}^{323} \frac{c dT}{T} = c \left[\ln \frac{323}{273} + \ln \frac{323}{523} \right] = c \ln \frac{323}{273}$$

= Same as (a)

$$\Delta S_{\text{Bath}}^1 + \Delta S_{\text{Bath}}^2 = -50 \text{ C} \left[\frac{1}{303} + \frac{1}{373} \right] = -0.155 \text{ C}$$

$$\Delta S_{\text{univ}} = 0.023 \text{ C}$$

(c) ΔS_{system} same

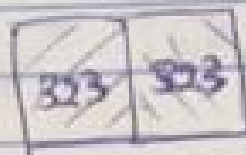
$$\Delta S_{\text{Bath}}^1 + \Delta S_{\text{Bath}}^2 + \Delta S_{\text{Bath}}^3 + \Delta S_{\text{Bath}}^4 = -25 \text{ C} \left[\frac{1}{298} + \frac{1}{323} + \frac{1}{242} + \frac{1}{273} \right]$$

$$= -0.580$$

$$\Delta S_{\text{univ}} = +0.02 \text{ C}$$

9.

C = constant



$$\Delta S = \int_{323}^{373} \frac{cdT}{T} + \int_{373}^{323} \frac{cdT}{T} = c \ln \frac{323}{373} + c \ln \frac{373}{323}$$

$$= c \ln \left(\frac{323^2}{323 \cdot 373} \right) = 0.024 C.$$

10.

$$\Delta S = \int \frac{cdT}{T} = \int \frac{aT + bT^3}{T} dT = \int (a + bT^2) dT$$

$$= aT + \frac{1}{3} bT^3.$$

16.

$$Q = 50 \text{ W}$$

$$Q = c \Delta T$$

$$\Delta T = \frac{50}{124} = 208$$

$$\Delta S = \int \frac{dq}{T} = c \ln \frac{T_f}{T_i} = 0.24 \ln \frac{508}{500} = 0.127 \text{ J}$$

18.

Maxwell relation #1

$$\left(\frac{\partial S}{\partial V} \right)_T = \left(\frac{\partial P}{\partial T} \right)_V$$

$$S = S(V, T)$$

$$dS = \left(\frac{\partial S}{\partial V} \right)_T dV + \left(\frac{\partial S}{\partial T} \right)_V dT = \left(\frac{\partial P}{\partial T} \right)_V dV + \frac{C_V}{T} dT$$

$$T dS = C_V dT + T \left(\frac{\partial P}{\partial T} \right)_V dV$$

18. Maxwell relation #2

$$\left(\frac{\partial V}{\partial T}\right)_P = -\left(\frac{\partial S}{\partial P}\right)_T$$

$$S(P, T) \quad dS = \left(\frac{\partial S}{\partial T}\right)_P dT + \left(\frac{\partial S}{\partial P}\right)_T dP$$

$$= \frac{C_p}{T} dT - \left(\frac{\partial V}{\partial T}\right)_P dP$$

$$T ds = C_p dT - \left(\frac{\partial V}{\partial T}\right)_P dP$$

$$du = T ds - P dv$$

$$\left(\frac{\partial u}{\partial v}\right)_T = T \left(\frac{\partial s}{\partial v}\right)_T - P$$

$$= T \left(\frac{\partial P}{\partial T}\right)_V - P$$

using Max
relation #

$$dU = Tds + \gamma dA$$

$$F = U - TS$$

$$dF = \gamma dA - SdT$$

$F(S, A)$ function of state

Consider
$$dF = \left(\frac{\partial F}{\partial S} \right)_A ds + \left(\frac{\partial F}{\partial A} \right)_S dA$$

$$\frac{\partial}{\partial A} \left(\frac{\partial F}{\partial T} \right)_A = - \frac{\partial S}{\partial A} = \frac{\partial}{\partial T} \left(\frac{\partial F}{\partial A} \right)_T = \frac{\partial \gamma}{\partial T}$$

Hence
$$\left(\frac{\partial S}{\partial A} \right)_T = - \left(\frac{\partial \gamma}{\partial T} \right)_A \quad ds = - \left(\frac{\partial \gamma}{\partial T} \right)_A dA$$

$$\therefore dU = -T \left(\frac{\partial \gamma}{\partial T} \right)_A dA + \gamma dA$$

$$= \left(\gamma - T \frac{\partial \gamma}{\partial T} \right)_A dA$$

$$U = \left[\gamma - T \left(\frac{\partial \gamma}{\partial T} \right)_A \right] A + \underline{\text{constant}}$$